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| <b>Must not be revealed</b> | <ol style="list-style-type: none"> <li>1. Basic personal information such as <b>author's name, gender, age, family relationship</b>, etc. <ul style="list-style-type: none"> <li>- Applicant's name should be covered anonymously.</li> <li>- Author information (<b>personal information, name</b>) listed at the top/bottom of the page according to the journal.</li> </ul> </li> <li>2. The name of the attached file should be changed as "published paper (1)" and the "published paper (2)". <ul style="list-style-type: none"> <li>- Be careful not to include personal information such as applicant's name in the name of the attached file.</li> </ul> </li> </ol> |
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|                             |  |
|-----------------------------|--|
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[Sample(Articles, Papers)]

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Carbon-titanium dioxide heterogeneous (photo)catalysts (C-TiO<sub>2</sub>) for highly efficient visible light photocatalytic application

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ABSTRACT

In this paper, a novel one-step method for the synthesis of a heterogeneous carbon-titanium dioxide (photo) catalyst (C-TiO<sub>2</sub>) is first reported. This synthesis method was performed at room temperature and atmospheric pressure using underwater plasma treatment for 15 min over various ratios of titanium and carbon sources. The resulting C-TiO<sub>2</sub> had anatase/brookite polycrystalline phases with turbostratic carbon and large surface areas. The bandgap energies were narrowed by the generation of reactive oxygen species and carbon bonds in the lattice of TiO<sub>2</sub>, extending optical absorption into the visible range. C60-TiO<sub>2</sub>, which had optimal ratios of carbon and TiO<sub>2</sub>, exhibited superior photocatalytic activities for methylene blue ( $k = 4.61 \text{ h}^{-1}$ ) under artificial solar irradiation due to its enhanced optical properties and numerous adsorption sites, which were approximately 10 times higher than those of commercial TiO<sub>2</sub> ( $k = 0.41 \text{ h}^{-1}$ ). This study represents a milestone of rapid and convenient methods to produce C-TiO<sub>2</sub> with high photocatalytic performance for environmental applications.

1. Introduction

As industry develops, the threat of environmental pollution is increasing. In particular, organic pollutants cause severe air and water pollution [1]. Numerous attempts to mitigate these organic pollutants have been documented; these attempts have included adsorption, electrical oxidation, and photocatalytic degradation [2–4]. Titanium dioxide (TiO<sub>2</sub>) is one of the strongest candidates as a feasible (photo)catalyst because of its various advantages, such as its low cost, nontoxicity, and ability to oxidize organic pollutants [5]. However, TiO<sub>2</sub> suffers from a wide bandgap, low activity under visible light, a high electron-hole recombination rate, and low adsorption ability, and these limitations strongly restrict its application in practical cases [6,7]. Many researchers have attempted to enhance the photocatalytic performance of TiO<sub>2</sub> using various approaches [6,8,9]. For example, TiO<sub>2</sub>-based heterogeneous (photo)catalysts, whose microstructures are modulated with

external impurities, have attracted increasing attention [10–14]. Among them, TiO<sub>2</sub> (photo)catalysts hybridized with carbon materials effectively enhance organic pollutant removal efficiency without using novel metal species [15]. The high conductivity of carbon materials may provide a path for photoexcited free electrons; therefore, charge-carrier separation occurs [12,14,16]. Additionally, the large surface areas of carbon materials improve the adsorption properties of organic pollutants [17,18]. In addition, hybridizing TiO<sub>2</sub> (photo)catalysts with carbon materials improves their light harvesting properties because white TiO<sub>2</sub> tends to reflect most irradiated light, but gray or black hybridized materials absorb more light in the visible and near-infrared regions [19–23].

TiO<sub>2</sub> (photo)catalysts hybridized with carbon materials are prepared using ultrasound radiation, ion exchange, and adsorption followed by hydrothermal treatment [24–27]. These methods usually require sol-gel processes that consume considerable time and energy [26,29];

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[Sample(Patents)]



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(54) 발명의 명칭 **코로나바이러스 감염증 COVID-19 치료용 펩타이드 및 이의 용도**


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**(57) 요약**

본 발명은 코로나바이러스 감염증 COVID-19 치료용 펩타이드 및 이의 용도에 관한 것으로서, 기존에 알려진 SARS-CoV의 RBD와 ACE2의 결합부위를 모사하는 펩타이드(P6)에 비해, 본 발명의 펩타이드는 SARS-CoV2 RBD의 새로운 에피토프와의 결합을 더욱 강하게 만들기 위해 아미노산을 구성하는 원자들 차원의 상호작용을 원천적으로 디자인한, 아미노산의 새로운 서열을 추가한 새로운 부분으로 구성되어 있다. 본 발명에서는, RBD와 hACE2 사이의 기존에 알려진 결합 경계면의 후면에 있는 D420, K458의 전하를 띤(Charged) 아미노산들과 추가적으로 상호작용할 수 있는, 확장된 펩타이드를 독창적으로 디자인하여 기존 알려진 펩타이드 보다 강하게 결합할 수 있는 새로운 디자인의 펩타이드를 제시하였으며, 본 발명의 펩타이드는 향후 COVID-19 치료제로서 높은 가능성을 나타내고 있다.

**대표도** - 도1

A



B

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PEP -----SEGRKTFLEPWREKELFYQSSLOKSEPR
PEPW NCSGLDGVYVITNCTNIMVQPNWEGKCTFLSEPWREKELFYQSSLOKSEPR
        
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